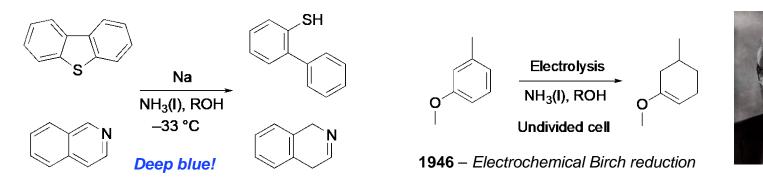


Metallic sodium reduction of heteroaromatics – 1944



Birch, A. J. 117. Reduction by Dissolving Metals. Part I. *J. Chem. Soc.* **1944**, No. 0, 430–436. https://doi.org/10.1039/JR9440000430.

Birch, A. Electrolytic Reduction in Liquid Ammonia.

Nature 1946, 158, 60. https://doi.org/10.1038/158060b0.

Hypothesis: Electrocatalysis for hydrogen-free bitumen upgrading?



Electrocatalysis Solvent, salts, RT

Undivided cell low overpotential *Reductive electrocatalysis* – HDS, HDN, HDO

Low sulfur, low nitrogen maltenes from asphaltenes

Oxidative electrocatalysis - Cold coking

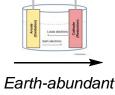
Mesophase asphaltenes, graphenes, quantum dots, reinforced carbon fibres



Hydrogen-free, ambient temperature upgrading to maltenes



Maltene-free asphaltenes New electrochemical extrusion

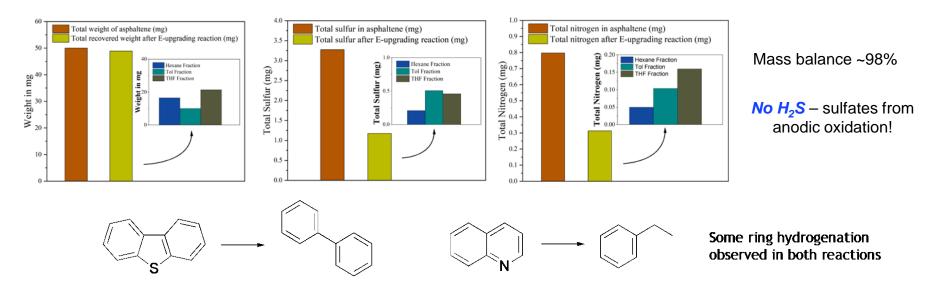


catalyst Low overpotential

Material	%N	%C	%Н	%S
Athabasca asphaltene	1.59	83.64	5.74	7.18
Maltenes post reaction	0.33	81.76	12.31	1.3
Asphaltenes post reaction	0.59	78.72	9.14	2.17

Optimal conditions give > 70% asphaltenes – sulfur and nitrogen below detection limits

Dramatic reduction in heteroatom content, high mass balance

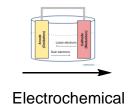




Asphaltene Upgrading

Mesophase synthesis at room temperature





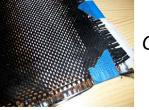


200 g asphaltenes; 4 L plastic bucket; electrodes purchased from Amazon.ca

90% pyridine insoluble (mesophase)

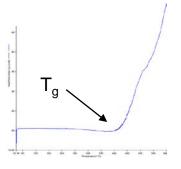
Mesophase luminescence (Quantum dots!)





Carbon fibres

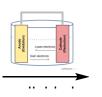
Removes occluded maltenes and isotropic asphaltenes

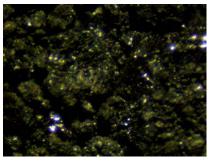


DSC (exothermic down)



InnoTech asphaltenes (mild heat-treated)





Polarized microscopy



+

+

Cold coking of asphaltenes – not much to look at....



Maltene-free asphaltenes (e-treated)



Electrochemical



Toluene solubles (recovered asphaltenes)

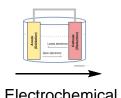


Petcoke (mostly pyridine insoluble)

Facile bitumen deasphalting



Whole bitumen



Electrochemical (w/special sauce)



Hexane solubles <10%



Toluene solubles >85%



Graphenes from black and and blue (hydrogen) cokes



Electrochemical (additives)

Amorphous carbon (3 anodes, 1cathode)



Spent anode (~12 h)

Low voltage, fast exfoliation (10-100X literature)



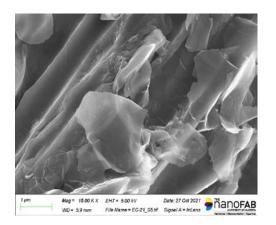
Graphenes

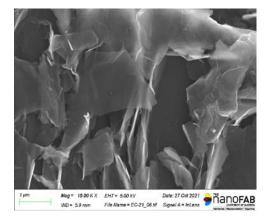
16 g of graphene (150 beaker, ON)

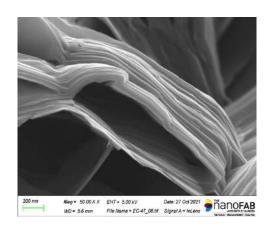
"Infinitely" scalable, parallel or series

Low oxygen; no acids (C/O = 9:1)

No post-processing deoxygenation



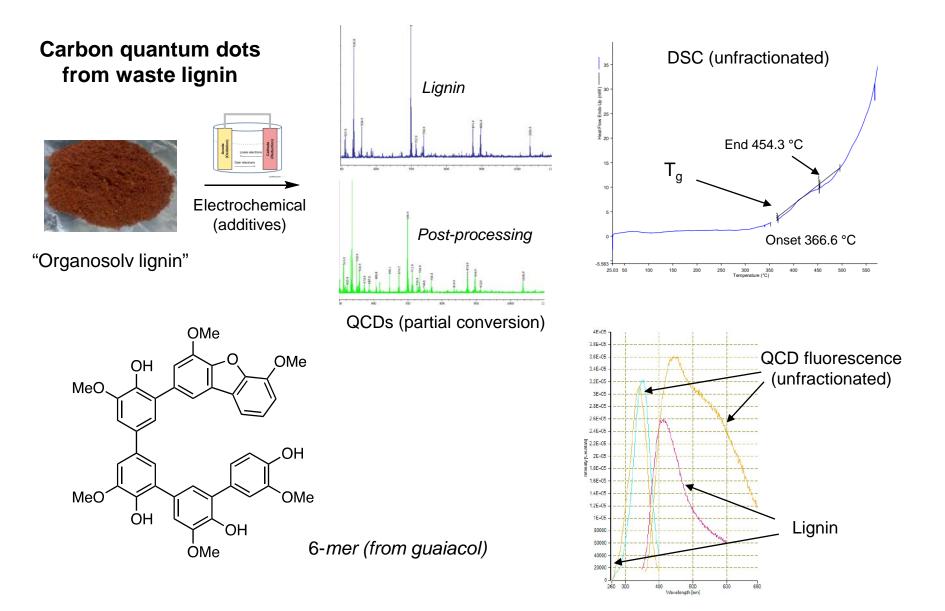




SEM imaging (UofA nanoFAB centre)

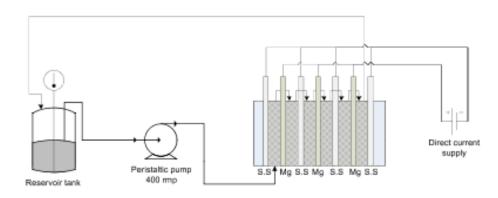


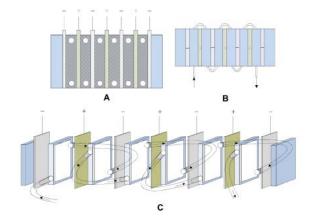
Value added products from lignin and cellulose waste

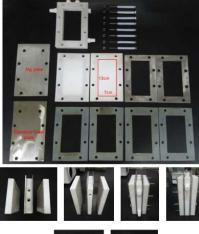


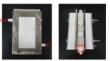


Engineering and scaling of flow electrocatalytic – Baran makes it look easy



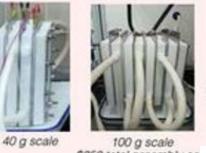






B Modular scalability of electrochemical Birch in flow





\$250 total assembly co

Modular Sustainable Cheap (3-D printed) Simple, robust Water friendly No clogging

Phil Baran et al. Scalable and Safe Synthetic Organic Electroreduction Inspired by Li-Ion Battery Chemistry. *Science* **2019**, *363* (6429), 838–845. https://doi.org/10.1126/science.aav5606.

Graphene-reinforced carbon fibres from Athabasca bitumen

UNIVERSITY OF





Dark materials matter!



Acknowledgements

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Collaborators:

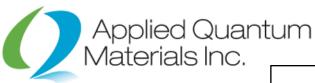
- Dr. Kevin Hodder (CME)
- Dr. Natalia Semagina (CME)



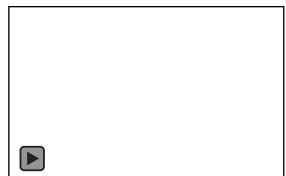


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FUTURE ENERGY SYSTEMS







Thanks, Murray!